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Unit 6 - Test (Chemical Compound) In chemistry, a compound refers to an entity that consists of two or more atoms, with at least two of those atoms from different chemical elements. These atoms associate via chemical bonds.

Unit 6 - Test(Chemical Compound) - ProProfs Quiz

The lines/curves indicate the maximum amount of solute which can be dissolved in 100g of water at each temperature. We call this a saturated solution.. Any amount above the curve is said to be supersaturated; there is more solute than the solvent can hold, therefore the excess precipitates out/settles to the bottom.. Any amount below the line is called unsaturated, meaning more solute can be ...

Unit 6 | chemistry

UNIT 6: RADIOACTIVITY AND NUCLEAR CHEMISTRY 5. A skeleton of a mammoth was analysed and found to contain 21.2% of the carbon-14 found in living bone. The half-life of carbon-14 is 5730 years. What is the most likely year of the mammoth ' s death? Use $\log(N/N_0) = -\lambda t$ $\lambda = \ln 2 / t_{1/2}$ $N/N_0 = 0.212$ $t = -\ln(0.212) / \lambda = 10021.2$ years; hence find t This year - t = year of death

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UNIT 6 – REDOX REACTIONS 6 • The oxidation number of an atom is the charge that would exist on an individual atom if the bonding were completely ionic • In simple ions, the oxidation number of the atom is the charge on the ion: - Na⁺, K⁺, H⁺ all have an oxidation number of +1 - Mg²⁺, Ca²⁺, Pb²⁺ all have an oxidation number of +2 - Cl⁻, Br⁻, I⁻ all have an oxidation number of -1

UNIT 6 REDOX REACTIONS - A-Level Chemistry

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Chemistry is the study of the composition, behaviour and properties of matter, and of the elements of the Earth and its atmosphere.

GCSE Chemistry (Single Science) - BBC Bitesize

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